Additions and Corrections

Electron Transfer. 114. Structure–Reactivity Relationships in the Redox Series Chromium(V,IV,III) and Vanadium(V,IV,III) [*J. Am. Chem. Soc.* **1993**, *115*, 3167–3173]. MANIK C. GHOSH* AND E. S. GOULD*

Due to a printing error, a number of superscript minus signs are not visible in the published paper.

In footnote 31, lines 2 and 3, the $Fe^{3+/2+}$ exchange rate is $10^{-3\pm1}$ M^{-1} s^{-1}. In footnote 38, line 4, k is equal to 2.6×10^{-3} s^{-1}. In footnote 39, line 2, the self-exchange rate estimated is $10^{-10.3}$ M^{-1} s^{-1}. In footnote 38, eq 10 is rate = $k[V^{III}][V^{IV}][H^+]^{-1}$.

Throughout the paper, the iridium complexes are $IrCl_6^{3-}$ and $IrCl_6^{2-}$, and the ligand ion is Lig^-.

The kinetic dimensions are the following: on p 3168, paragraph 3, line 8, M^{-1} cm^{-1}; in Table V, footnote *d*, M^{-1} s^{-1}; in footnote 25, line 4, M^{-2} s^{-2}; and in footnote 31, line 5, M^{-1} s^{-1}.

Generation, Some Synthetic Uses, and 1,2-Vinyl Rearrangements of Secondary and Tertiary Homoallyllithiums, Including Ring Contractions and A Ring Expansion. Remarkable Acceleration of the Rearrangement by an Oxyanionic Group [*J. Am. Chem. Soc.* **1993**, *115*, 3855–3865]. BOGUSLAW MUDRYK AND THEODORE COHEN*

In the printing process, Figures 1 and 2 (top and bottom, respectively) both have their main features obscured. The correct figures are as follows.